

The effect of clinoptilolite on ^{137}Cs binding in broiler chickens

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The objective of this study was to evaluate the ^{137}Cs binding capacity of clinoptilolite. In the first *in vitro* experiment we investigated sorption of ^{137}Cs to natural and modified forms of clinoptilolite in highly acid solution, prepared to be similar to that of the gut of pigs (pH = 2–3) at 37°C. In the second *in vivo* experiment ^{137}Cs binding to a modified form of clinoptilolite was studied in orally contaminated broiler chickens. ^{137}Cs sorption in the high acidity solution depended on clinoptilolite concentration and varied between 30–85% of the initial activity. In the chickens, three hours after ^{137}Cs administration, there was 67% and 63% lower accumulation of ^{137}Cs in meat and edible organs (respectively) and seven hours after ^{137}Cs administration, there was 69% and 49% lower accumulation of ^{137}Cs in meat and edible organs (respectively) compared to the controls with no clinoptilolite added in food. Natural and modified forms of clinoptilolite have been shown to high sorption efficiency towards ^{137}Cs ions and could be recommended as possible radiocaesium binders in domestic animals.

Key words: clinoptilolite, ^{137}Cs , sorption, broilers

Introduction

Contamination of the environment with radioactive substances has become one of the most important problems in modern civilization. Therefore, there is a need for constant and systematic control of radionuclide concentrations in the biosphere. Radioactive contamination of

the environment as a consequence of the Chernobyl nuclear accident has led to contamination of vegetation and feedstuffs for animal nutrition with biologically significant radionuclides, ^{90}Sr , ^{131}I and ^{137}Cs . Prevention of contamination via the oral ingestion of these radionuclides is a priority in radioprotection. In animal production, there is a need to ensure that contamination of animal products is below intervention levels.

Howard et al. (2001) reviewed the current availability of countermeasures for preventing or reducing contamination of animal products by radioisotopes of iodine, caesium and strontium.

Reducing contamination via the gut of domestic animals and the consequent health risk for humans, requires the development of methods which use binding substances to bind radio-caesium in the digestive tract of animals and eliminate it through faeces. After the first nuclear accidents, initial investigations were conducted on ¹³⁷Cs sorption. Potassium was added into the feed to stimulate radiocaesium elimination from the organism (Williams and Patrick 1957). Beresford et al. (1989) used bentonite as a radiocaesium binder in sheep and achieved a reduction of between 20–80%. Currently, the best caesium binding effects have been achieved by Prussian Blue compounds, the most commonly used compound being AFCF - ammonium ferric hexacyanoferrate (Arnaud et al. 1988, Hove and Ekern 1988, Unsworth et al. 1989, Hove and Hansen 1993, Paasikallio et al. 2000). Reduction of 90–98% was achieved in cows with 3 g of AFCF daily, calves and pigs 2 g of AFCF daily (Giese 1988) and in broilers (Vitorovic 1993) with 0.2 g of AFCF daily per chick. Philippo et al. (1988), Pethes et al. (1988) and Jandl and Novosad (1995) demonstrated that zeolite (a clinoptilolite clay mineral) can be used as a possible binder for ¹³⁷Cs in the gut of domestic animals.

Zeolites are crystalline, hydrated aluminosilicates that have an indefinite three-dimensional structure. Their characteristic is the ability to exchange water and some of its constitutional cations without significant changes in their structure. Natural zeolites are used as catalysts in many reactions, as molecular sieves and for cation exchange (Ming and Mumpton 1989). Considering this, zeolites have wide use as animal feed additives with the ability for mycotoxine adsorption (Rodrigues 1991, Harvey et al. 1993, Amon and Dobesic 1994, Krilov et al. 1994). Clinoptilolite could be also used in agriculture to reduce ¹³⁷Cs deposition in soil because of its

ability to bind ¹³⁷Cs (Paul and Jones 1995, Campbell and Davies 1997, Jones et al. 1999).

The objective of the present study was to evaluate the ¹³⁷Cs binding capacity of a natural white form of clinoptilolite and different modified clinoptilolite forms in *in vitro* and *in vivo* conditions.

Material and methods

In the *in vitro* experiment, natural clinoptilolite from Zlatokop mine was used. The investigated zeolite had above 80% clinoptilolite mainly in Ca(II) form. Silicate analysis was used for Si and Al determination, atomic absorption for Na, K, Ca, Mg determination and the Kjeldhal method for NH₄⁺ ion. The cation exchange (Na, K, Ca and Mg) capacity was above 130 cmol kg⁻¹. In the first stage of this experiment, 100 mg and 300 mg of clinoptilolite were treated with 100 ml electrolytic solution of HCl, NaCl and KCl, to which 1440 Bq ¹³⁷Cs was added. The acidity of the solutions was initially adjusted to be pH 2–3, with NaOH, which resembles the acidity condition within the gut of pigs. Sorption was performed in a water bath at 37°C with slow shaking. Contact times were 2, 4, 6 and 8 hours and three replicates were taken for each period of time. After sorption, the samples were centrifuged and remaining activity measured in the separated liquid fraction. In the second stage of this experiment, we compared ¹³⁷Cs binding capacity of natural and mono-ionic, sodium (Na) and ammonium (NH₄) forms of clinoptilolite. The hydrated samples of natural clinoptilolite were treated with 3M NH₄Cl and 3M NaCl solutions to prepare the mono-ionic forms of clinoptilolite. 500 mg of the sample was equilibrated with 50 ml electrolytic solution to which 1440 Bq ¹³⁷Cs was added under the same solution composition and acidity as in the first stage of the experiment. The contact time was 6 hours at 37°C after which the remaining activity of ¹³⁷Cs in solution was measured, as before.

For the *in vivo* experiment, broiler chickens (six weeks of age) were fed with their normal diet and each chick was orally given a 3 ml CsCl solution with a total activity of 3120 Bq. In addition, the experimental group was given a 6 ml aqueous solution of clinoptilolite modified with NaCl called Minazel (1.6 g Minazel per chick which is approximately 1–1.5% in the diet). Principles of laboratory animal keeping were in accordance with national laws for animal protection and welfare. Birds of each group were sacrificed 3 and 7 hours (five chicks per group) after the contamination was administered. Samples of total musculature and edible organs (heart, gizzard and liver) were taken from each chick. A HP Ge detector with high-energy resolution (1.85 KeV – 1.33 MeV) was used for gamma spectrometry measurement of activity of the samples. The measurement time was 40,000 s. The effect of clinoptilolite treatments on ¹³⁷Cs binding was evaluated by analysis of variance and differences between treatments were tested using t-test.

Results and Discussion

Administration of the lower clinoptilolite concentration (100 mg) reduced by half the initial ¹³⁷Cs activities (52%) in solution during 6 hours

of incubation. Prolongation of the contact time gave no further decreasing in solution activity (48.6% for 8 hours). Using a higher clinoptilolite amount (300 mg), the equilibrium of Cs ions was also established after 6 hours, but 75% (74.7% in 6 hours and 74.2% in 8 hours) of initial ¹³⁷Cs activities was sorbed. The higher clinoptilolite administration therefore have a higher sorption efficiency (between 67.6–74.7%) than that of the lower amount (30.4–48.6%). The activities of solution depending of different contact times and clinoptilolite concentrations are given in Table 1.

Natural and prepared forms of clinoptilolite (Na- and NH₄- clinoptilolites, 500 mg) showed similar and significant effects on ¹³⁷Cs sorbtion, sorbing 84–85% of initial activity (Table 2). The higher ¹³⁷Cs sorption efficiency of clinoptilolite in this experiment compared to the results in the first stage was the result of using higher clinoptilolite concentrations. Philippo et al. (1988) established that clinoptilolite and hexacyanocobalt ferrate are the most effective adsorbers of radiocaesium *in vitro*. When used *in vivo* they significantly reduced the absorption of radiocaesium by sheep fed contaminated herbage.

In our *in vivo* experiment, ¹³⁷Cs binding efficiency was evaluated for the modified clinoptilolite form - Minazel, in broiler chickens. Three hours after administering ¹³⁷Cs contamination and the mineral, significantly (P < 0.01) lower ¹³⁷Cs activity concentration was obtained in the

Table 1. Clinoptilolite efficiency in sorption of Cs¹³⁷ from a solution, with an activity of 1440 Bq.

Time of contact (hours)	¹³⁷ Cs activity in solution (Bq)				Sorbed activity (% of starting activity)	
	100 mg clinoptilolite		300 mg clinoptilolite		100 mg clinoptilolite	300 mg clinoptilolite
	Mean	SD	Mean	SD		
2	1001.0	15.0	466.0**	10.0	30.4	67.6
4	894.0	4.0	383.0**	8.0	37.8	78.4
6	748.0	15.0	364.0**	11.0	48.0	74.7
8	739.0	12.0	371.0**	10.0	48.6	74.2

SD = standard deviation

** = statistically significant differences (P < 0.01) between treatment of clinoptilolite concentration.

Vitorovic, G. et al. The effect of clinoptilolite on ^{137}Cs binding in broiler chickens

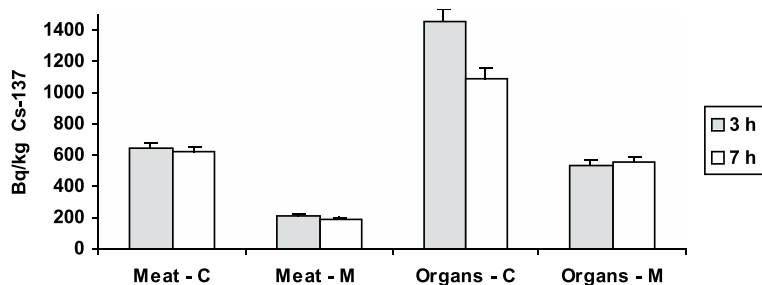


Fig. 1. Efficiency of Minazel in reducing ^{137}Cs accumulation in the meat and edible organs of broiler chickens, orally contaminated with 3150 Bq ^{137}Cs per chick, abbreviation C is for “control” and M for “Minazel” (means and standard errors).

meat of broilers treated with Minazel than in the control group. Compared to the control group, the reduction of ^{137}Cs accumulation in the treated group was 67%. Seven hours after contamination and mineral administration, the decrease due to ^{137}Cs binder was 69%. Pethes et al. (1988) found that accumulation of radiocaesium was significantly decreased by adding 6% zeolite to broiler feed. The accumulation of ^{137}Cs in zeolite treated groups ranged 62–70% of the non-treated groups. Jandl and Novosad (1995) investigated *in vivo* reduction of radiocaesium in sheep using modified clinoptilolite (ZEOFe) and non-modified clinoptilolite (ZEO). More than ten times lower sorption efficiency has been observed with ZEO comparison to ZEOFe. In our investigation, the modified clinoptilolite Minazel, showed significant protection effect in edible organs (heart, gizzard and liver) of broilers contaminated with ^{137}Cs . Three hours after contamination and mineral administration there was about 63% lower accumulation of ^{137}Cs in edible organs of the group treated with Minazel

compared to controls; after seven hours it was 49%.

Considering that these are preliminary results in our investigation of the ability of clinoptilolite to bind ^{137}Cs in the digestive organs of domestic animals, it is still too early to give suggestions for its practical application and to compare it with other ^{137}Cs binders (bentonite, vermiculite, AFCF). It is also too early to evaluate animal products obtained using zeolite as ^{137}Cs binder in human nutrition especially because in this investigation the degree of oral ^{137}Cs contamination of chickens (3150 Bq per day) was far above potential radioactive contamination of chick feeds in accidental situations (like the one produced by Chernobyl). It was used to obtain a clear data on the efficiency of clinoptilolite in ^{137}Cs binding in the digestive tract of chickens. In our further investigations more attention will be given to radioactive contamination of animal feeds in practical conditions with long-term observations in the ability of clinoptilolite to bind ^{137}Cs in orally contaminated animals.

Table 2. Efficiency of different forms of clinoptilolite in ^{137}Cs sorption from a solution, with an activity of 1440 Bq.

Forms of clinoptilolite	^{137}Cs activity (Bq)		Sorbed activity (% of starting activity)
	Mean	SD	
Natural	230.0	11.0	84.0
Sodium (Na-)	215.0	11.0	85.1
Ammonium (NH_4^-)	215.0	7.0	85.3

SD = standard deviation

Conclusion

Natural clinoptilolite and modified forms of clinoptilolite have a high sorption efficiency towards ^{137}Cs ions. In Serbia, there are deposits of this particular mineral of high quality. Because of that, and its low price, it could be recommended as a possible radiocaesium binder in domes-

tic animals. Additional *in vivo* studies are needed to evaluate optimal clinoptilolite concentrations for different species, breed and categories of domestic animals.

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