

Long time-series of chemical and isotopic compositions of Vesuvius fumaroles: evidence for deep and shallow processes

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ABSTRACT

Long time-series of chemical and isotopic compositions of Vesuvius fumaroles were acquired in the framework of the volcanic surveillance in the 1998-2010 period. These allow the identification of processes that occur at shallow levels in the hydrothermal system, and variations that are induced by deep changes in volcanic activity. Partial condensation processes of fumarolic water under near-discharge conditions can explain the annual ¹⁸O and deuterium variabilities that are observed at Vesuvius fumaroles. Significant variations in the chemical compositions of fumaroles occurred over the 1999-2002 period, which accompanied the seismic crisis of autumn 1999, when Vesuvius was affected by the most energetic earthquakes of its last quiescence period. A continuous increase in the relative concentrations of CO₂ and He and a general decrease in the CH₄ concentrations are interpreted as the consequence of an increment in the relative amount of magmatic fluids in the hydrothermal system. Gas equilibria support this hypothesis, showing a P_{CO₂} peak that culminated in 2002, increasing from values of ~40 bar in 1998 to ~55-60 bar in 2001-2002. We propose that the seismic crisis of 1999 marked the arrival of the magmatic fluids into the hydrothermal system, which caused the observed geochemical variations that started in 1999 and culminated in 2002.

1. Introduction

The present volcanic-hydrothermal activity at Vesuvius volcano is relatively low level. The main evidence of this activity is: (a) widespread fumarolic emissions that are accompanied by diffuse soil CO₂ degassing in the crater area [Chiodini et al. 2001, Frondini et al. 2004]; (b) CO₂-rich groundwaters along the southern flank of Vesuvius and in the adjacent plain [Caliro et al. 1998, Federico et al. 2002, Caliro et al. 2005]; and (c) seismic activity with epicenters clustered inside the crater [Saccorotti et al. 2002, Del Pezzo et al. 2004].

The most notable seismic activity was registered on October 1999. This crisis included an earthquake of M_L 3.6, which was the highest magnitude recorded for at least 25 years, and possibly since the last eruption of Vesuvius in 1944 [Zollo et al. 2002, Del Pezzo et al. 2004]. After the October

1999 seismic crisis, seismic activity decreased to low levels, with shallow hypocenters mainly clustered within the volcanic edifice [Del Pezzo et al. 2004].

The degassing area of the Vesuvius crater is characterized by the presence of fumarolic vents that are sited on the crater rim and at the bottom of the crater (Figure 1). Fumarolic fluids discharged by these fumaroles on the crater rim are of relatively low temperatures (<75 °C) and are mainly composed of atmospheric components. Fumaroles from the crater bottom have a composition that shows H₂O and CO₂ as the major components, followed by H₂, H₂S, N₂, CH₄, CO and He (in order of decreasing content), and discharge temperature of about 95 °C, i.e. the condensation temperature of fumarolic fluids at the crater altitude (P_{atm} = 0.91 bar).

The presence of significant CH₄ and NH₃ contents provides evidence to the origin of these fluids being from a hydrothermal environment [Chiodini et al. 2001]. In particular, on the basis of a comprehensive geochemical study of crater fumarolic fluids collected in the period 1998-1999, Chiodini et al. [2001] suggested the presence of a high-temperature hydrothermal system that is located below the Vesuvius crater.

The aim of this study is to investigate the eventual changes that have affected this hydrothermal system over a relatively long period (1998-2010), on the basis of the data acquired in the framework of volcanic surveillance. These data include chemical compositions of the rim fumaroles, chemical and isotopic compositions of the crater-bottom fumaroles, and soil CO₂ flux data relative to a continuous monitoring station that is located at the bottom of the Vesuvius crater (Figure 1).

We show modifications of the fumarolic fluids due to the occurrence of seasonal and deep processes. In particular, we focus on the variations that accompanied and followed the seismic crisis of 1999, which caused detectable anomalies in the Vesuvius groundwaters [Federico et al. 2004, Madonia et al. 2008].

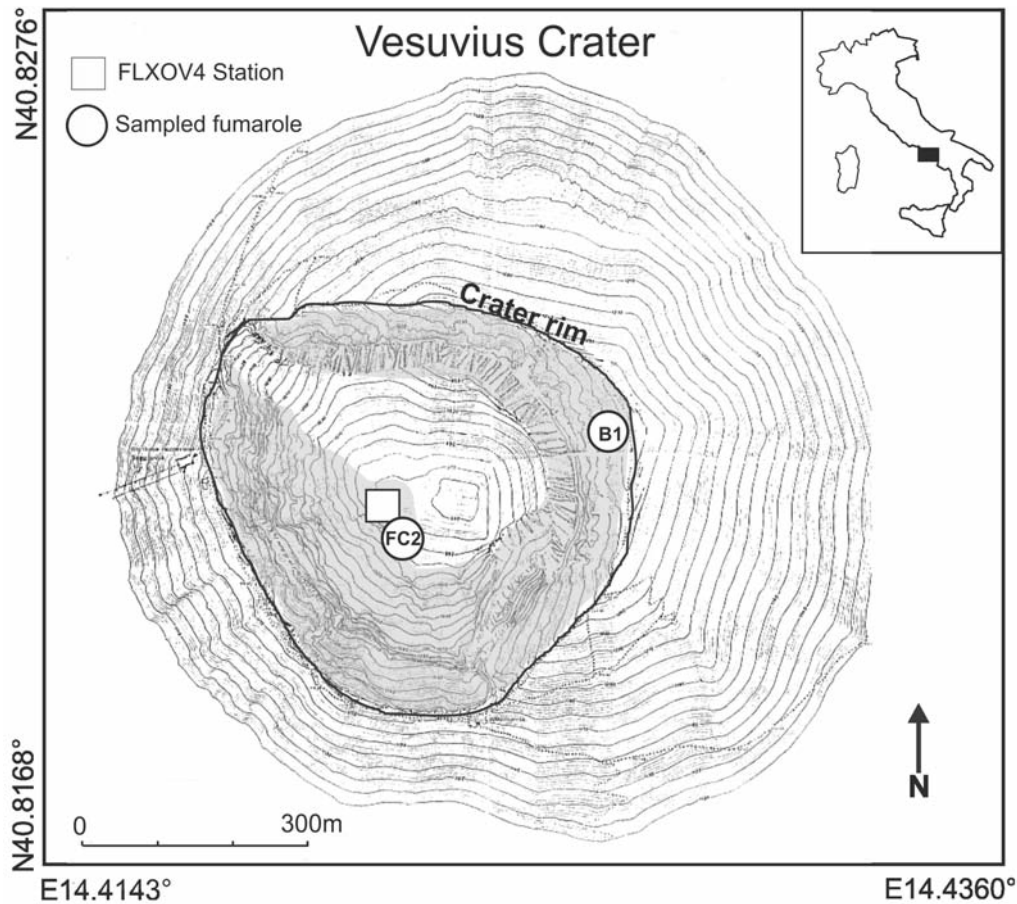


Figure 1. Sketch map of the Vesuvius crater, and location of the sampled fumaroles and the soil CO₂ flux station (FLXOV4). Inset: Location of Vesuvius volcano within Italy.

1.1. Volcanological setting

The Somma-Vesuvius volcanic complex lies in the southern sector of the Campanian Plain, a Plio-Quaternary structural depression that extends from the western part of the Apennine chain to the eastern coast of the Tyrrhenian Sea. The latter was generated by an important subsidence of Mesozoic carbonate rocks, and was subsequently filled by Quaternary sedimentary and volcanic deposits that have reached a thickness of about 2,000 m beneath the volcanic edifice [Aprile and Ortolani 1979]. Somma-Vesuvius was formed at the intersection of two regional fault systems with NW–SE and NE–SW stress directions [Ippolito et al. 1973, Pescatore and Sgrosso 1973]. The volcanic complex consists of the relatively old strato–volcano, Monte Somma, that underwent a series of caldera collapses that culminated in 18 ka B.P. [Cioni et al. 1999, Ventura et al. 1999], and a younger intracaldera cone, Vesuvius. The Somma-Vesuvius activity started between 18 ka and 37 ka [Principe et al. 1999], and it was characterized by great variability in eruptive styles: relatively frequent, open-conduit activity, that consisted of effusive eruptions that were sometimes accompanied by Strombolian events, and that alternated with less frequent Plinian and sub-Plinian eruptions that typically followed long periods of quiescence [Arnò et al. 1987]. The Somma-Vesuvius volcanic products generally

show alkali-potassic compositions and large ranges of silica contents, from nearly saturated rock (leucite-basalts, leucite-trachybasalts, latites, and trachytes) to highly under-saturated rock (leucititic tephrites, leucititic tephrit-phonolites, and leucititic phonolites) [Joron et al. 1987]. Effusive and Strombolian events discharged small batches (<0.001 km³) of primitive magma, whereas Plinian eruptions involved larger batches (0.1 to >1.0 km³) of more evolved magma. These magmas differentiated into reservoirs that are located within a thick sequence of chiefly carbonate rock at depths of 3–5 km b.s.l. [Barberi and Leoni 1980, Barberi et al. 1981, Santacroce 1983]. After the last Plinian eruption, which took place in 1631 [Rosi et al. 1993], Vesuvius was marked by open-conduit activity, which ended with the eruption of 1944. Since then, the volcano has been characterized by fumarolic activity.

1.2. The conceptual geochemical model of the Vesuvius hydrothermal system

The crater rim fumaroles are characterized by low temperatures (62–76 °C) and they discharge a mixture of air (46%–72%), steam (25%–45%) and CO₂ (0.2%–2%). The air, which in all of the samples is the main component of the fumaroles, is most probably included in fluids in the upper part of the fumarolic conduits that are located in the very

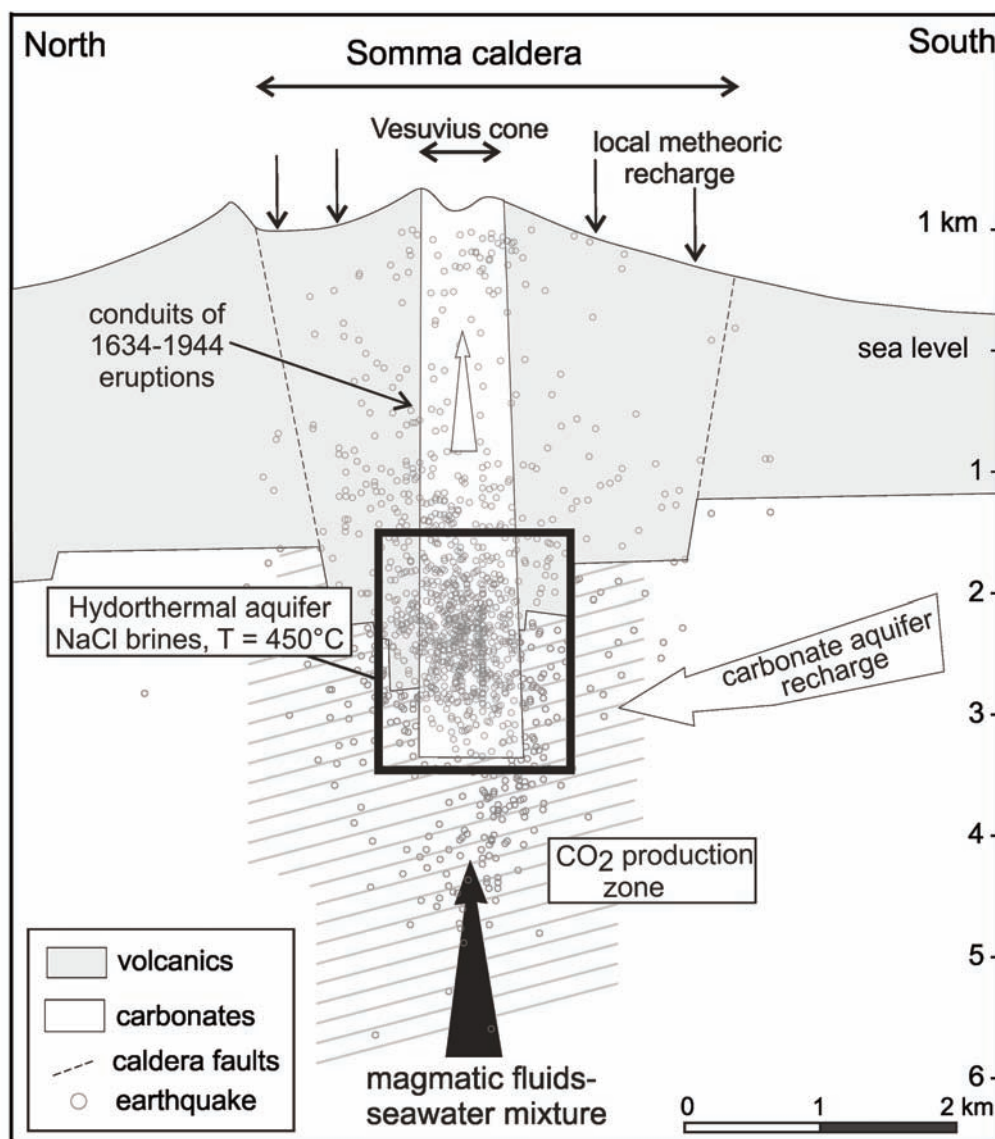


Figure 2. Conceptual geochemical model of the volcanic hydrothermal system of Vesuvius [after Chiodini et al. 2001], refined with the data from the present study (see Section 3.3.1). The seismic events registered in the 1995-1999 period [Saccorotti et al. 2002] are reported for comparison.

permeable products of the Vesuvius cone. Because of this heavy air contamination, these fumaroles are not indicative of the deep system and they have not been considered for the elaboration of the conceptual geochemical model. In contrast, the crater bottom fumaroles discharge fluids that are representative of the deep systems: H_2O and CO_2 are the major components, followed by H_2 , H_2S , N_2 , CH_4 , CO and He (in order of decreasing content). Excluding minor possible contamination during the sampling, air is absent (N_2/Ar generally >200). The discharge temperatures, of about 95°C , are close to the saturation temperature of steam for the fumarolic fluids at the crater altitude. Fumarolic gases do not show any detectable SO_2 , HCl , and HF , due to the scrubbing of magmatic gases within the hydrothermal system.

In the following we report and discuss the conceptual geochemical model of these fumaroles (Figure 2) that was elaborated by Chiodini et al. [2001], and that is here assumed as the reference model for the interpretation of the chemical

and isotopic variations that have been observed over the last 10 years of monitoring.

The origin of the main components, H_2O and CO_2 , were investigated on the basis of isotopic data. The interpretation of stable isotopes of water was, however, not sufficiently constrained by the data, possibly because of the occurrence of secondary processes. Fumarolic water was thus interpreted as either meteoric water enriched in ^{18}O through water-rock oxygen isotope exchange, or a mixture of meteoric and arc-type magmatic water [Chiodini et al. 2001].

The carbon isotopic signature of the fumarolic CO_2 was compatible, at least partially, with derivation from metamorphic reactions that involve marine carbonates and that take place in the thick carbonate sequence located at depths >2.5 km underneath Vesuvius [Chiodini et al. 2001]. More recently, this partial derivation of CO_2 from decarbonation processes was confirmed by petrological considerations regarding the process of carbonate assimilation

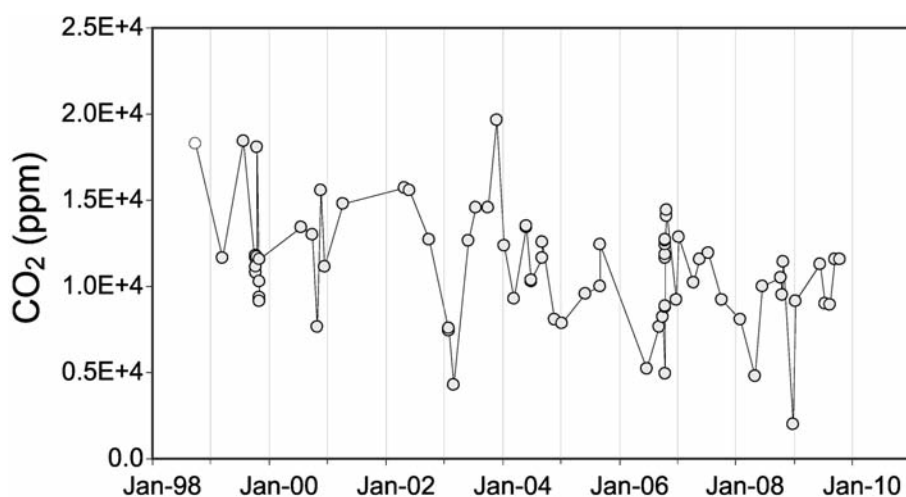


Figure 3. Chronogram of the CO₂ contents of the B1 rim fumarole during the 1998-2010 period.

by the Vesuvius magma [Iacono Marziano et al. 2009].

To derive indications of the temperature–pressure conditions of the hydrothermal system, Chiodini et al. [2001] considered gas equilibria in the CO₂-CH₄-CO-H₂-H₂O gas system in the presence of NaCl brines. In particular, $f_{\text{H}_2\text{O}}$, the fugacity of water, and the vapor–liquid distribution coefficients B_i were considered to be controlled by the presence of either 3 m NaCl solutions or halite saturated brines. For the FC2 fumarole, i.e. one that is regularly monitored and is considered in the present study, equilibrium temperatures were 445 ± 9 °C considering 3 m NaCl solutions, and 429 ± 8 °C for the halite-saturated brine. Such high equilibration temperatures were confirmed by the individual CO/CO₂, H₂/H₂O, and CH₄/CO₂ ratios, by the H₂-H₂S-H₂O and H₂-N₂-NH₃-H₂O gas equilibria, and by the H₂-Ar geoinicator [Chiodini et al. 2001]. In addition, similar high temperatures (450-500 °C) were confirmed on the basis of the C-isotopic exchange between CH₄ and CO₂ [Fiebig et al. 2004], leaving little uncertainty of the thermal state of the Vesuvius hydrothermal system. Due to the uncertainty of the salt content of the hydrothermal liquid, the estimation of total fluid equilibrium pressure ($P_{\text{tot}} = P_{\text{H}_2\text{O}} + P_{\text{CO}_2}$) varies from 260 bar, in the case of halite saturation, to 480 bar, for 3 m NaCl solutions [Chiodini et al. 2001]. Considering a hydrostatic regime, the deepest hydrothermal reservoir might be located at depths of roughly 2.5 km to 5.0 km, within the carbonate sequence, which is present at depths >2.5 km underneath Vesuvius [Berrino et al. 1998]. It is worth noting that the estimated depths of the deepest hydrothermal reservoir roughly coincide with the major frequency distribution of the hypocenter of earthquakes recorded in the 1995-1999 period [Saccorotti et al. 2002], suggesting a possible relationship between these. In particular, Saccorotti et al. [2002] proposed that the seismic activity at Vesuvius is triggered by diffusion of a pressure front, which is possibly associated with the degassing process of the hydrothermal system.

2. Materials and methods

This study is based on the compositional data of fumaroles B1 (crater rim) and FC2 (crater bottom) the locations of which are shown in the map in Figure 1. Although there are other fumarolic vents in the crater area that have been analyzed occasionally, we have limited the discussion here to the FC2 and B1 fumaroles, for which a continuous time-series of chemical and isotopic compositions is available.

In particular, we used 88 chemical analyses, and analyzed 72 oxygen isotopic compositions, and 61 hydrogen isotopic compositions of the steam condensate of the FC2 fumarolic vent, and 79 chemical analyses of B1 fumarole, which were collected periodically from 1998 to 2010 in the framework of volcanic surveillance of the Campanian volcanoes.

Gas samples for the determination of the chemical compositions of FC2 fumarole were collected in 200 ml glass vials under-vacuum, which were equipped with a Teflon stopcock fixed with a rubber O-ring, which contained a 4 N NaOH solution [Giggenbach 1975, Giggenbach and Gouguel 1989]. Dry gas samples were collected for determination of CO concentrations. Condensates of water vapor and non-condensable gases were sampled by flowing the fumarolic gases through a condenser cooled at 20 °C to 30 °C by water. The chemical compositions of the gases and the isotopic compositions of the CO₂ were analyzed at the geochemistry laboratory of INGV-Osservatorio Vesuviano. The gas constituents were analyzed following the methods of Cioni and Corazza [1981], as modified for the analysis of He, Ar, O₂, N₂, H₂ and CH₄. The chemical compositions of these nonabsorbed gases that were mainly present in the headspace over the NaOH solution were measured by gas chromatography, through a unique injection on two molecular-sieve columns (MS 5Å capillary, 30 m × 0.53 mm × 50 μm; He and Ar were the carrier gases) using TCD detectors. The carbon dioxide and sulfur species absorbed in the alkaline

solution were analyzed after oxidation via H_2O_2 , and by acid-base titration and ion chromatography, respectively (analytical error, $\pm 3\%$). Because of the reaction in alkaline solution to form $COOH^-$ [Giggenbach and Matsuo 1991], CO was analyzed for dry gas samples, using gas-chromatography separation with a MS 5Å capillary ($30\text{ m} \times 0.53\text{ mm} \times 50\text{ }\mu\text{m}$); He was used as carrier gas, using a TCD detector.

Some isotopic analyses (data from 1996 to 2002) of oxygen and hydrogen (water) were performed at the Geochemistry Laboratory of INGV-Palermo, using a Finnigan Delta plus mass spectrometer. Oxygen isotopic compositions were measured after equilibration with CO_2 at $25\text{ }^\circ\text{C}$ [Epstein and Mayeda 1953], H was measured after reduction with Zn [Coleman et al. 1982], and C after cryogenic purification of CO_2 (analytical errors: $\delta D \pm 1\text{‰}$, $\delta^{18}O \pm 0.1\text{‰}$ and $\delta^{13}C \pm 0.1\text{‰}$). Isotopic data since 2003 were performed at the Geochemistry Laboratory of INGV-Osservatorio Vesuviano, using a Finnigan Delta plusXP continuous flow mass spectrometer coupled with a GasbenchII device (analytical errors: $\delta D \pm 1\text{‰}$, $\delta^{18}O \pm 0.08\text{‰}$ and $\delta^{13}C \pm 0.06\text{‰}$).

Gas samples for the determination of the chemical compositions of B1 fumarole were collected in 20 mL glass vials that were equipped with a Teflon stopcock fixed with a rubber O-ring, by flowing the fumarolic gases through a condenser cooled at $20\text{ }^\circ\text{C}$ to $30\text{ }^\circ\text{C}$ with water. The dry gas was analyzed by gas chromatography, through a unique injection on two columns (Molecular Sieve 5 Å capillary $30\text{ m} \times 0.53\text{ mm} \times 50\text{ }\mu\text{m}$, and Hysep Q packed $90' \times 1/4'$) using He as the carrier gas on the columns and TCD detectors.

The analytical data are reported in the electronic Annex 1.

3. Time-series of the chemical and isotopic compositions of the fumaroles during 1998-2010

3.1. Time-series of the rim fumarole B1

The B1 fumarole is fed by air that is variably contaminated by CO_2 of deep origin, as suggested by some of the data on the carbon isotopic composition ($\delta^{13}C_{CO_2} \sim -0.7\text{‰}$) that are close to the values measured in the crater bottom fumaroles (generally from -0.5‰ to 0.7‰) that are not affected by air contamination. The CO_2 concentration is thus representative of the relative contribution of the deep component. The chronogram of Figure 3 shows that over time the CO_2 content of this fumarole is scattered, probably because the amount of air that enters the fumarolic conduits is controlled by variable meteorological conditions, such as the wind and atmospheric pressure. In spite of this dispersion that is caused by secondary processes, the chronogram shows a general decreasing trend in CO_2 , which can be interpreted as a decrease in the time of the flux of the deep component.

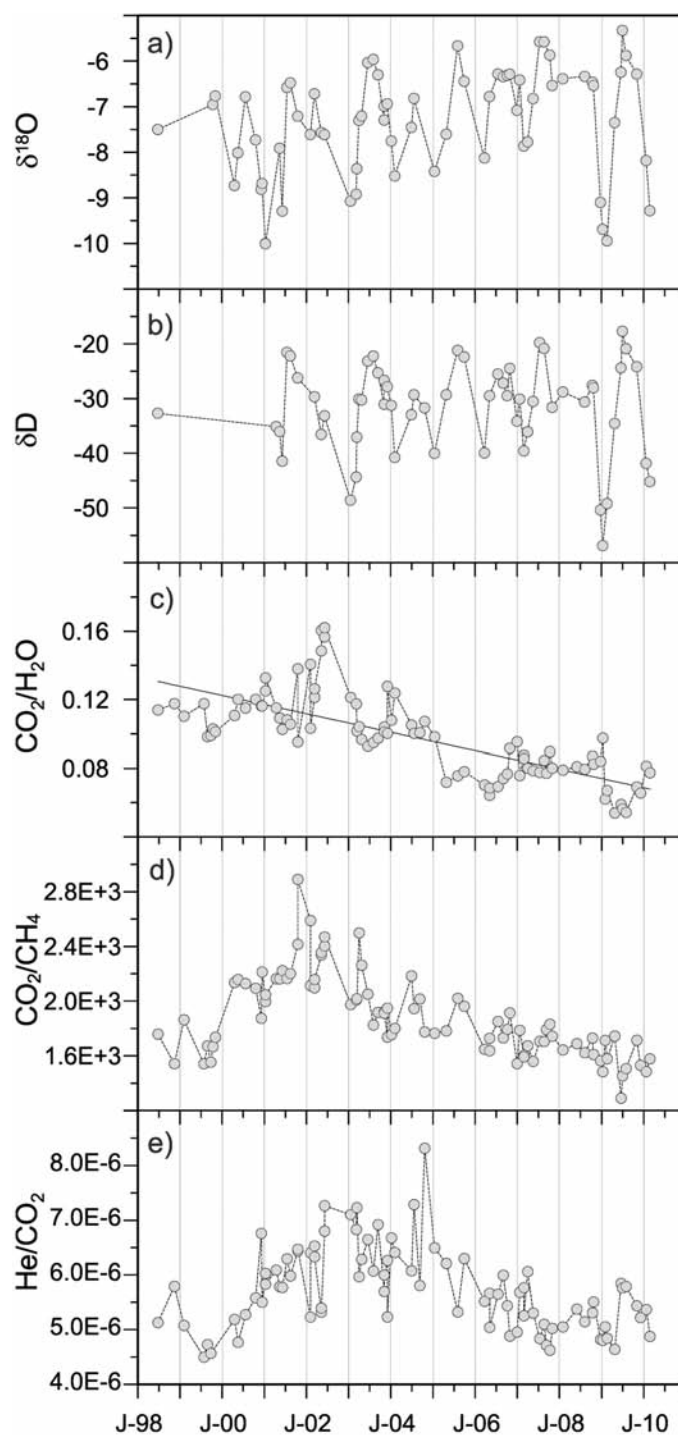


Figure 4. Time-series of selected parameters: a) $\delta^{18}O_{H_2O}$; b) δD_{H_2O} ; c) CO_2/H_2O ; d) CO_2/CH_4 ; and e) He/CO_2 measured at fumarole FC2 over the 1998-2010 period. The isotopic compositions of the fumarolic water (4a and 4b) are characterized by an annual cycle, while the long-period variations affect the CO_2/H_2O , CO_2/CH_4 and He/CO_2 ratios characterized by a unique peak in 2002.

3.2. Time-series of the FC2 crater bottom fumarole

Time-series of selected parameters ($\delta^{18}O_{H_2O}$, δD_{H_2O} , CO_2/H_2O , CO_2/CH_4 , He/CO_2) that were measured at the FC2 fumaroles during 1998-2010 are reported in the chronograms of Figure 4. Figure 4 highlights the occurrence of compositional variations of different frequencies: short-period variations are typical of the stable isotopes of the

condensates, which show an evident annual cycle (Figure 4a,b), while the $\text{CO}_2/\text{H}_2\text{O}$, CO_2/CH_4 and He/CO_2 ratios are affected by long-period variations that are characterized by a unique peak value, which is positive for He and CO_2 and negative for CH_4 , and which occurred in 2002 (Figure 4c-e).

This annual cycle that characterizes the water isotopic composition testify to the occurrence of shallow processes that are linked in some way to seasonal effects, while the behavior of the gas species might be caused by some deep variations in the system that feeds the fumaroles.

3.2.1. Shallow processes

The $\delta^{18}\text{O}$ values of steam condensates range from -10‰ to -5‰ , while their δD values range from -57‰ to -18‰ , with both showing a marked annual cycle (Figure 4a,b). This annual cycle is more evident through examination of Figure 5, where the values are plotted against the months of sampling. Both the oxygen and hydrogen isotopic compositions show sinusoidal trends that are characterized by minimum values from December to February, while the maxima are reached from June to August, indicating strong seasonal control of the isotopic composition of the fumarolic water.

A possible process that might cause such seasonal variations is seen in the partial condensation of the fluids within the fumarolic conduits, a process that in winter will be enhanced by the lower ambient temperatures, and/or by the infiltration of cold meteoric water and melted snow. This condensation is a plausible process, because both the relatively low flow rate of the fumarole and its discharge temperature (mean, 94.2 °C) are close to the boiling temperature (94.5 °C , at this specific altitude). During

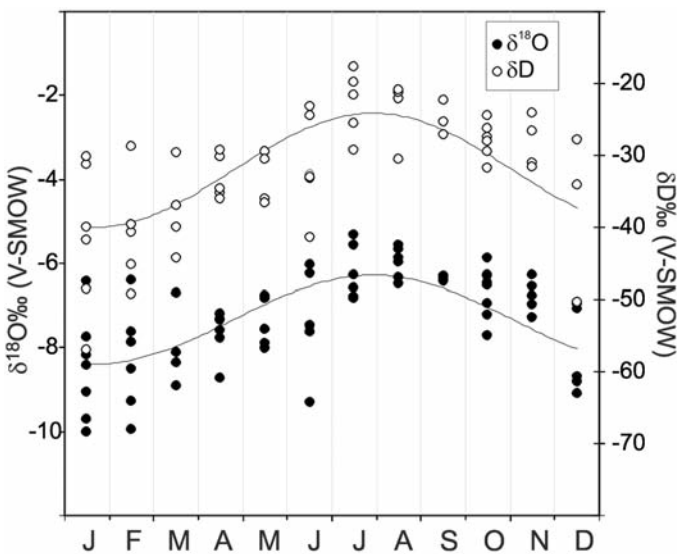


Figure 5. $\delta^{18}\text{O}$ and δD values of steam condensates versus the months of sampling. More depleted samples (in heavy isotopes) characterize the wintry months, indicating strong seasonal control of the isotopic composition of fumarolic water.

condensation, the isotopic fractionation causes the formation of a liquid that is enriched in heavy isotopes (^{18}O and deuterium [D]), which leaves a residual vapor of a lighter isotopic composition. This process explains why during the winter season, when condensation is enhanced, the fumarole discharge fluids are characterized by lighter isotopic compositions. These effects of condensation have been modeled using two different approaches.

The first approach (model A) considers only the fractionation due to the condensation of the steam at boiling point. In more detail, we computed the δD , $\delta^{18}\text{O}$ compositions of the vapor remaining after condensation (the residual vapor) for different fractions of condensation, in a multi-step process. Considering the fractionation factors between liquid and vapor at 94.5 °C ($1000\ln\alpha_{l-v} = 30.14$ and 5.30 for D and ^{18}O) [Horita and Wesolowski 1994], the isotopic compositions are computed at each step by:

$$\delta\text{D}_o = (1 - y) \times \delta\text{D}_r + y \times (\delta\text{D}_r + 1000\ln\alpha_{l-v}) \quad (1)$$

$$\delta\text{D}_r = \delta\text{D}_o - (y \times 1000\ln\alpha_{l-v}) \quad (2)$$

$$\delta^{18}\text{O}_o = (1 - y) \times \delta^{18}\text{O}_r + y \times (\delta^{18}\text{O}_r + 1000\ln\alpha_{l-v}) \quad (3)$$

$$\delta^{18}\text{O}_r = \delta^{18}\text{O}_o - (y \times 1000\ln\alpha_{l-v}) \quad (4)$$

where the subscripts *o* and *r* refer to the original and the residual vapors, and *y* is the fraction of steam condensed at each step.

In this exercise, the starting vapor was selected as that of the fumarolic composition characterized by a heavier isotopic composition (sample of 22/08/2007: $\delta^{18}\text{O} = -5.58\text{‰}$, $\delta\text{D} = -20.84\text{‰}$), which is considered as representative of a noncondensed sample. Although, this assumption is not entirely constrained by the data, the possible choice of a more positive isotopic composition for the original fluid would be purely speculative. The theoretical compositions of the residual vapor, i.e. of the vapor remaining after condensation, are reported in Figure 6, as the "model A" line. This model does not explain the measured compositions that are systematically enriched in ^{18}O , with respect to the condensation line.

This systematic deviation is explained by the contemporary occurrence of oxygen exchange between CO_2 and steam during the condensation. Chiodini et al. [2000] demonstrated that in fumarolic gas this process is fast enough to allow rapid isotopic re-equilibration of fumarolic fluids at their discharge temperatures, within a wide temperature range ($100\text{--}1,000\text{ °C}$). In particular, 52 samples where the oxygen isotopic compositions in both H_2O and CO_2 were measured show $\delta^{18}\text{O}_{\text{CO}_2} - \delta^{18}\text{O}_{\text{H}_2\text{O}}$ differences ($36\text{‰} \pm 1\text{‰}$) that are close to the theoretical fractionation at 95 °C (34.1‰).

For this reason, the second model (Figure 6, "model B") takes into account the occurrence of oxygen exchange

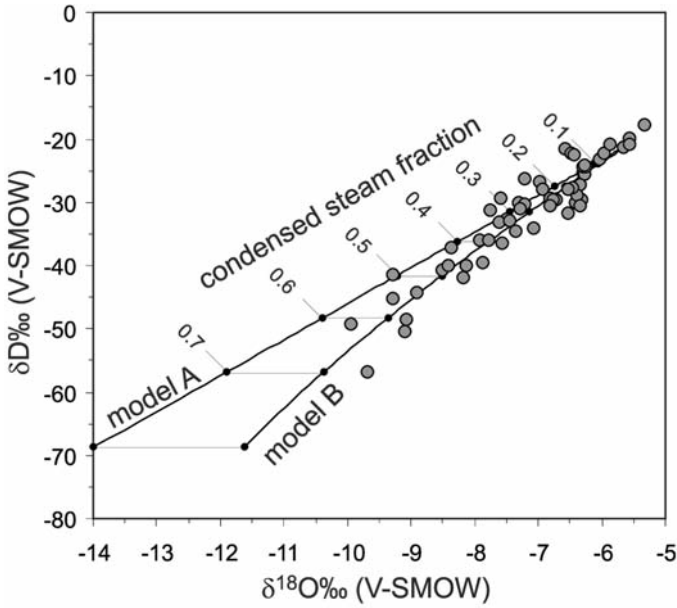


Figure 6. Plot of δD vs. $\delta^{18}O$ of the fumarolic condensates (fumarole FC2). Condensation models under near discharge conditions ($94.5^\circ C$) are also reported, with the parent fluid sample of 22/08/2007. Model A refers to simply multistep steam condensation, while model B refers to multistep steam condensation that takes into account the oxygen isotopic exchange between CO_2 and water vapor during the process. The good fitting of the analytical and theoretical (model B) data support the possible occurrence of this process.

between CO_2 and steam during the condensation. Practically, this more complex condensation model is based on the solution of the isotopic mass-balance for the oxygen in the $H_2O + CO_2$ system, which can be written as:

$$\delta^{18}O_{tot} = (\delta^{18}O_r + 1000\ln\alpha_{CO_2-v}) \times \chi_{CO_2-v} + \delta^{18}O_r \times \chi_v + (\delta^{18}O_r + 1000\ln\alpha_{1-v}) \times \chi_l \quad (5)$$

$$\delta^{18}O_r = \delta^{18}O_{tot} - (1000\ln\alpha_{CO_2-v} \times \chi_{CO_2-v} + 1000\ln\alpha_{1-v} \times \chi_l) \quad (6)$$

where $\delta^{18}O_{tot}$ is the oxygen isotopic composition in the $CO_2 + H_2O$ system before the condensation, χ_{CO_2} , χ_v and χ_l are the oxygen atom fractions for CO_2 , vapor and liquid water in the $CO_2 + H_2O$ system, and $1000\ln\alpha_{CO_2-v} = 34.20$ is the CO_2 - H_2O_{vapour} oxygen fractionation factor computed at $94.5^\circ C$ using the equation of Richet et al. [1977]. The oxygen atom fractions are given by:

$$\chi_{CO_2} = 2X_{CO_2} / (2X_{CO_2} + X_{H_2O}) \quad (7)$$

$$\chi_l = y \times X_{H_2O} / (2X_{CO_2} + X_{H_2O}) \quad (8)$$

$$\chi_v = (1 - y) \times X_{H_2O} / (2X_{CO_2} + X_{H_2O}) \quad (9)$$

The relatively good fit of the measured points with model B suggests that the annual ^{18}O and D variability observed at the Vesuvius fumaroles is mainly due to condensation processes, which are more efficient during the cold seasons. The fraction of condensed steam can be

roughly evaluated at 0.3–0.4, by comparing the theoretical and measured data.

In principle, the same trend of the fumarolic δD - $\delta^{18}O$ values can be explained by other more complex processes, such as mixing of the deep hydrothermal fluids (either vapor or liquid) with cold groundwaters, formation of mixed parent liquids, and separation of equilibrated vapors [Giggenbach and Stewart 1982, Henley and Stewart 1983, Chiodini et al. 2001, Federico et al. 2010]. These processes possibly occur in peripheral areas of the hydrothermal systems, where meteoric waters can interact with the deep fluids, causing the formation of mixtures at lower temperatures, with respect to the undiluted original fluids. These processes are unlikely to occur at Vesuvius, where gas equilibria indicate almost constant and high temperature and pressure values. In contrast, the simplest condensation model that is proposed here for the first time is based on processes that are both directly observable at the fumarole (i.e. condensation) and demonstrated by analytical measurements (i.e. isotopic equilibrium between oxygen in water and in CO_2 ; see above).

The removal of water from fumarolic fluids by condensation implies a concurrent increase in the relative concentration of the incondensable gas, which should show an annual cycle similar to that of the water isotopes. In this case, however, the occurrence of evident long-term variations, as highlighted in Figure 4c-e, can hide the eventual annual variations. For example, the CO_2 content of the fumaroles shows a clear decreasing trend from 1998 to 2010. We tried to de-trend the CO_2 chronogram by adding the residual of the linear regression of CO_2 versus time to the mean CO_2 value (see Figure 4c). The de-trended values, which are plotted against the months of sampling in Figure 7, show an annual

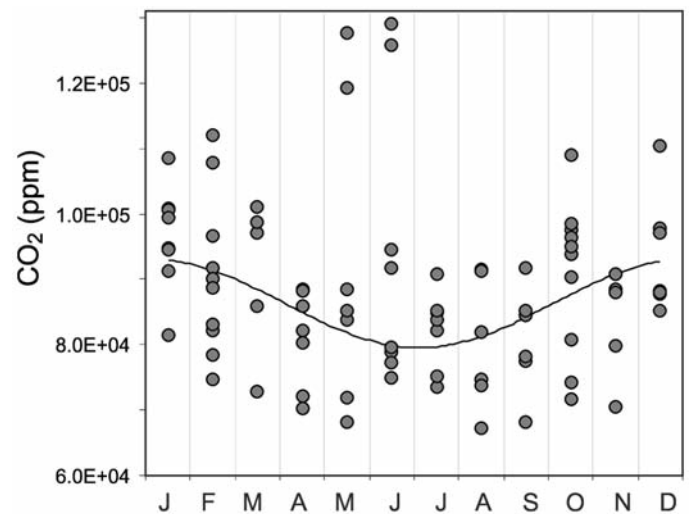


Figure 7. CO_2 de-trended contents of the FC2 fluids (see text) versus the months of sampling. A cyclical annual behavior is seen, with the higher values characterizing the cold months (enhanced condensation), which supports the occurrence of condensation processes.

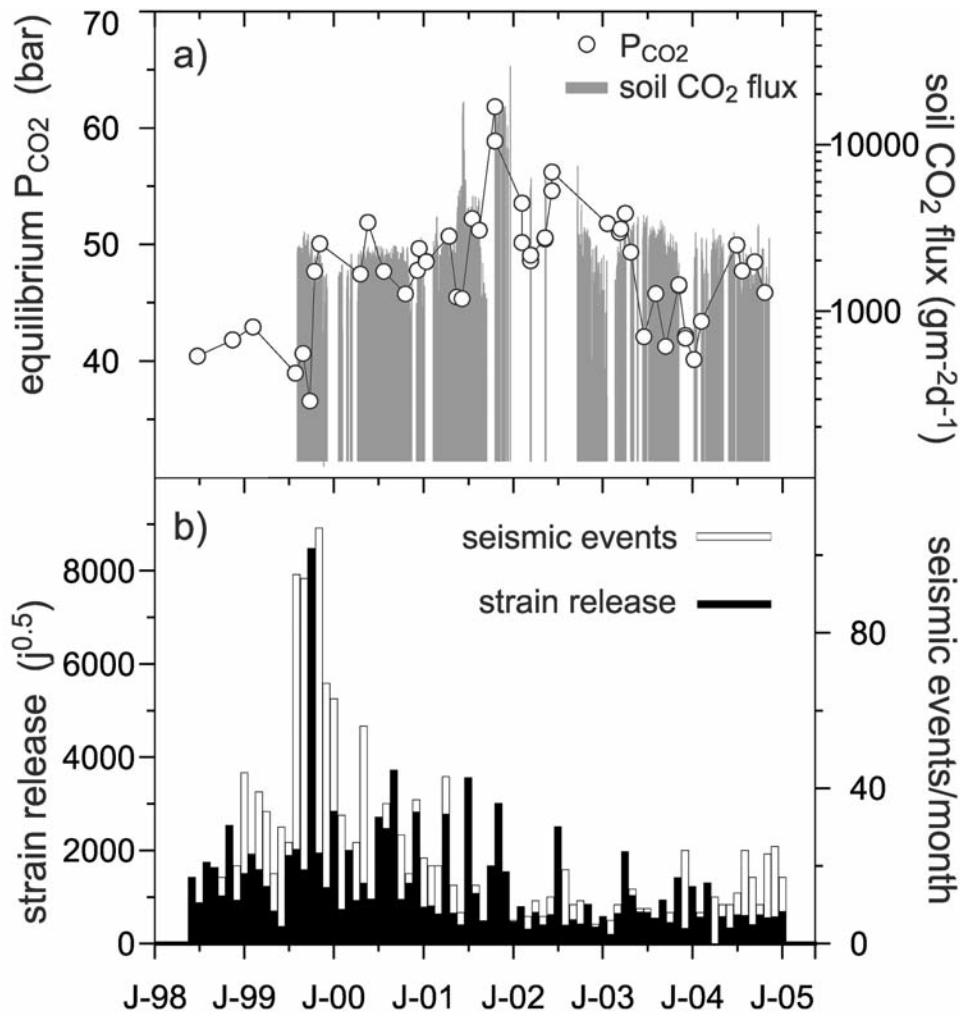


Figure 8. Time variations of the equilibrium P_{CO_2} and the soil CO_2 flux (a), and of the seismic activity reported both as number of events per month and as the strain release [Giudicepietro et al. 2010] (b). A P_{CO_2} peak starts in concomitance with the seismic crisis of October 1999, passing from values of ~ 40 bar in 1998 to ~ 55 – 60 bar in 2001–2002, in good agreement with the CO_2 flux data recorded by the automatic FLXOV4 station that increased by one order of magnitude.

cycle that, in agreement with the isotope variations, has the highest concentrations in the winter season (90,000–110,000 ppm) and the lowest in the summer (70,000–90,000 ppm), which confirms the occurrence of condensation that is driven by seasonal processes. Only a few points clearly deviate from the annual trend. These refer to samples collected in 2002, when other parameters showed anomalous behaviors due to the arrival at the surface of a deeply generated geochemical signal.

3.2.2. Deep processes

In addition to the shallow secondary processes discussed above, other evidence highlights that the Vesuvius fumaroles have been affected by compositional changes caused by deep processes. An evident positive peak culminated in 2002 characterizes both the CO_2/CH_4 and the He/CO_2 ratios (Figure 4d,e). In particular, the CO_2/CH_4 ratio increased sharply from a value of $\sim 1,500$ in 1998 to $\sim 3,000$ in 2002, and then gently decreased to the original values. It is worth noting that CH_4 is a gas species that differentiates hydrothermal gases, where it is present at relatively high

concentrations, from magmatic fluids, where it is normally absent or present at very low concentrations. Due to this large difference between magmatic and hydrothermal fluids, the increase in the CO_2/CH_4 ratio provides a powerful geochemical indicator of the arrival of hot, oxidizing, and possibly magmatic, fluids at the shallow levels of volcanoes characterized by the presence of hydrothermal systems above the magmatic ones [Chiodini 2009]. The arrival of deeper fluids, which is possibly characterized by the relatively high He concentrations, also explains the increase in the He/CO_2 ratio that occurred in 1999–2002, concurrent with the increases in the CO_2/CH_4 ratio (Figure 4e).

In the case of Vesuvius, CH_4 is formed in the hottest zones of the hydrothermal system by CO_2 reduction, as suggested by both chemical and isotopic gas equilibria [Chiodini et al. 2001, Fiebig et al. 2004]. Because of the relatively slow kinetics of the CO_2 to CH_4 reduction, the arrival of a pulse of deeper and hotter fluids that are poor in CH_4 would cause a rapid increase in the CO_2/CH_4 ratio, followed by a decrease that reflects the slow re-establishment of the original hydrothermal conditions. On the other hand,

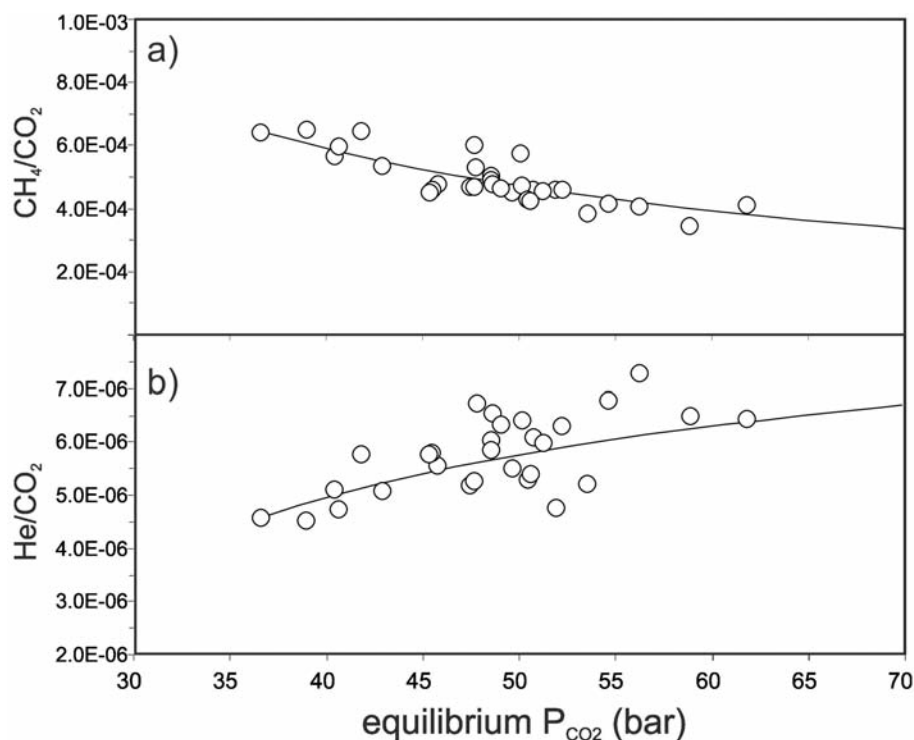


Figure 9. CH_4/CO_2 (a) and He/CO_2 (b) ratios of fumarolic fluids versus equilibrium P_{CO_2} . Lines refer to a dilution model of fumarolic fluids with a hypothetical magmatic gas characterized by the absence of CH_4 and by $\text{He}/\text{CO}_2 = 9 \times 10^{-6}$. The good fitting of data supports the concept that a pulse of deep, possibly magmatic, fluids caused the 2002 anomaly.

the other gas species that are sensitive to redox conditions (i.e. H_2 and CO) would quickly re-adjust to the new conditions, as these species are characterized by fast kinetics [Chiodini et al. 1993, Giggenbach 1987]. Assuming the presence of halite saturated brines, H_2 and CO were used as indicators of the P_{CO_2} of the hydrothermal system of Vesuvius [Chiodini et al. 2001]. To better investigate the origin of the 2002 anomaly, we computed the equilibrium P_{CO_2} during the period 1998-2005, adopting the same approach as Chiodini et al. [2001]. The chronogram reported in Figure 8a shows that the 2002 He- CH_4 - CO_2 anomaly (Figure 4c-e) was also accompanied by an evident increase in the equilibrium P_{CO_2} , which went from ~ 40 bar in 1998 to ~ 55 -60 bar in 2001-2002. It is worth noting that these P_{CO_2} values are in good agreement with the CO_2 flux data, an extensive and independent parameter that is recorded by the automatic station located in the bottom of the Vesuvius crater, near fumarole FC2 (Figure 1, FLXOV4). During 2002, the CO_2 flux increased by a factor of ~ 1 order of magnitude, going from $\sim 1,000 \text{ gm}^{-2}\text{d}^{-1}$ in 1998-1999 to $\sim 10,000 \text{ gm}^{-2}\text{d}^{-1}$ in 2002 (Figure 8a).

To sum up, all of the observed variations are in agreement with the arrival of a pulse of deep, possibly magmatic, fluids rich in CO_2 and He and poor in CH_4 . To further validate this hypothesis, a simple mass-balance dilution model of fumarolic fluids with a hypothetical magmatic gas is presented in Figure 9, where the CH_4/CO_2 and He/CO_2 values of the 1998-2002 samples are plotted against the P_{CO_2} estimated on the basis of the H_2 and CO

contents. In Figure 9, the lines represent the values expected for the injection into a box, which represents the hydrothermal system theoretically, for a gas with $\text{CH}_4/\text{CO}_2 = 0$ (no CH_4) and $\text{He}/\text{CO}_2 = 9 \times 10^{-6}$. Practically, we considered that the initial gas composition was that of the lowest P_{CO_2} sample (September 1999: $P_{\text{CO}_2} = 36.6$ bar). Consequently, we computed the initial P_{CH_4} and P_{He} on the basis of their composition ($P_{\text{CH}_4} = P_{\text{CO}_2} \times \text{CH}_4/\text{CO}_2 = 2.35 \times 10^{-2}$ bar; $P_{\text{He}} = P_{\text{CO}_2} \times \text{He}/\text{CO}_2 = 1.67 \times 10^{-4}$ bar). This simple model can account for the compositional variations of the fumarolic fluids, again supporting the concept that a pulse of deep, possibly magmatic, fluids caused the 2002 anomaly.

It is worth noting that the anomaly starts in concomitance with the seismic crisis of Autumn 1999 (Figure 8b). This crisis included the main earthquake recorded at Vesuvius after the last eruption of 1944 (the event of October 9, 1999; M_L 3.6; depth ~ 3 km beneath the central cone) [Del Pezzo et al. 2004]. In our opinion, the earthquakes of 1999 marked the arrival of the magmatic fluids into the hydrothermal system, which caused the above-described variations that culminated in 2002. In agreement with our interpretation, Saccorotti et al. [2002] proposed that at least a part of the seismic activity at Vesuvius is triggered by diffusion of a pressure front, which is possibly associated with the degassing process of the hydrothermal system. The seismic activity reported in Figure 8b (both as number of events per month and as monthly strain release) shows an intense period of activity starting from the October 1999, when there were the highest values, which then remained at considerable levels to the end of 2002.

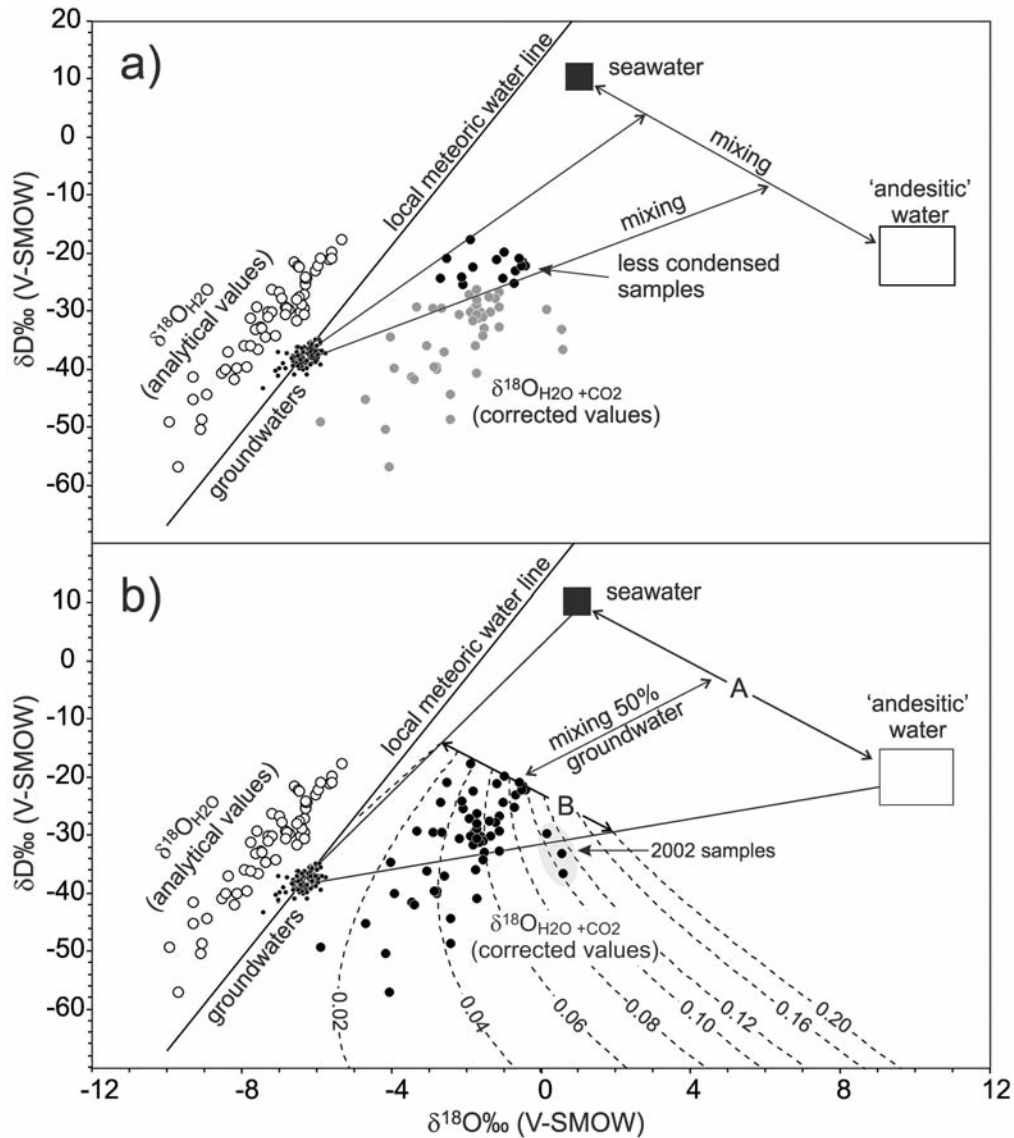


Figure 10. a) δD versus $\delta^{18}O$ diagram. Analytical isotopic compositions of fumarolic condensates and the computed $\delta^{18}O_{tot}$ values of the $H_2O + CO_2$ system under discharged conditions are reported, together with the local meteoric water line [Caliro et al. 1998], isotopic composition of the groundwaters circulating in the volcanic aquifers of Vesuvius [Caliro et al. 2005] and other possible end-members of mixing (seawater and andesitic waters). The fumarolic data were more or less unaffected by condensation processes (i.e. more positive isotopic values, black dots) and they support an origin of the water due to a mixture of andesitic water, seawater and local meteoric water. b) Mixing-condensation model involving seawater, andesitic water and local groundwater. The assumed compositions of the end members are: seawater $\delta D = 10\text{‰}$, $\delta^{18}O = 1\text{‰}$, $X_{CO_2} = 0$; andesitic water $\delta D = -20\text{‰}$, $\delta^{18}O = 10\text{‰}$, $X_{CO_2} = 0.4$; groundwaters: $\delta D = -38\text{‰}$, $\delta^{18}O = -6.3\text{‰}$, $X_{CO_2} = 0$. The mixing line A represents the vapor generated by seawater–andesitic water mixtures. The mixing line B represents a dilution of line A by 50% in groundwater. Dashed lines represent the condensation patterns of line B vapors. The number of dashed lines refer to the different CO_2 molar fractions of the pre-condensed line B vapors.

4. Origin of the fumarolic water

The large dataset of water isotopic composition that was acquired during the 1998–2010 period of volcanic monitoring can be used to refine the interpretation of the origin of the fumarolic water, with respect to previous interpretations [Chiodini et al. 2001]. For this purpose, we computed the oxygen isotopic composition of the $H_2O + CO_2$ system for the fumarolic fluids under discharge conditions, by means of the following equation:

$$\delta^{18}O_{tot} = (\delta^{18}O_v + 1000 \ln \alpha_{CO_2-v}) \times \chi_{CO_2} + \delta^{18}O_v \times (1 - \chi_{CO_2}) \quad (10)$$

where symbols are the same as for Equation (5).

The computed $\delta^{18}O_{tot}$ data plot in Figure 10a on the right side of the local meteoric water forming a large cluster of points. Considering only the samples that were less affected by condensation (Figure 10a, black dots), Figure 10a suggests a derivation of fumarolic water due to variable mixing of three end-members: local meteoric water, arc-type magmatic water, and seawater. It is worth noting that this isotopic evidence on the involvement of seawater in the Vesuvius hydrothermal system is a novel finding, which is consistent with the already recognized presence of saline brines in this hydrothermal system [Chiodini et al. 2001].

To explain the highly scattered distribution of the samples, in Figure 10b the data are compared with the composition expected for a possible mixing–condensation model. In particular, the grid in Figure 10b represents the vapor generated by seawater–andesitic water mixtures (Figure 10b, line A) which are diluted by 50% in the groundwater (Figure 10b, line B). The resulting vapors are subsequently condensed (Figure 10b, dashed lines). The effects of the condensation are strongly affected by the CO₂ concentrations of the pre-condensed mixtures.

Even if the selected composition for the magmatic component is largely unconstrained, as is the fraction of groundwater, on the whole, the model reproduces well the large variability of the measured isotopic composition. It is worth noting that the samples that would be more rich in the andesitic water component are those of 2002, i.e. the same samples where independent observations highlighted the presence of larger fractions of magmatic fluids (see above).

5. Conclusions

Long time-series of chemical and isotopic compositions of the Vesuvius fumaroles have allowed us to identify processes that occur with different frequencies: short-period variations appear to be linked to processes that occur at shallow levels in the hydrothermal system, while relatively long time variations are induced by deep changes in the volcanic activity.

The isotopic compositions of the fumarolic waters show evident annual cycles (Figure 5), with the more depleted values seen in the autumn-to-winter period. Taking into account the oxygen exchange between CO₂ and steam, condensation processes at near discharge conditions can explain the annual ¹⁸O and D variabilities observed at the Vesuvius fumaroles. These processes are more efficient during the cold seasons, when the ambient temperatures are lower. The fraction of condensed steam can be roughly evaluated by comparing the theoretical and measured data, as 0.3 to 0.4.

Taking into account the condensation processes, the analysis of the large dataset of water isotopic composition that was acquired during the 1998-2010 period of volcanic monitoring has allowed the refinement of the interpretation for the source of fumarolic water. Considering the isotopic oxygen exchange between CO₂ and water vapor, the isotopic data suggest a derivation of the water from a mixture of meteoric, arc-type magmatic water, and seawater. The isotopic evidence of seawater is a novel finding that agrees with the presences of high saline brines in the hydrothermal system of Vesuvius [Chiodini et al. 2001].

Superimposed on the seasonal effects, the time series of the chemical composition of fumaroles reveals that important variations in the activity of Vesuvius occurred in

the 1999-2002 period. A continuous increase in the CO₂ and He relative concentrations and a general decrease in the CH₄ concentrations are interpreted to be the consequences of the increase in the relative amount of magmatic fluids in the hydrothermal system, which are rich in CO₂ and He, and poor in CH₄. Gas equilibria support this hypothesis, showing a P_{CO₂} peak that culminated in 2002, passing from values of ~40 bar in 1998 to values of ~55–60 bar in 2001-2002 (Figure 8a). It is worth noting that the P_{CO₂} values are supported by the CO₂ soil-flux data that were recorded by the automatic station FLXOV4, which is located close to fumarole FC2 (Figure 1); these increased concurrently by about one order of magnitude (Figure 8a). These changes appear to be related to the seismic crisis that culminated with the main earthquake recorded on October 9, 1999, of M_L 3.6 [Del Pezzo et al. 2004]. This event had the highest magnitude that has been recorded for at least 25 years, and possibly since the last eruption of Vesuvius (1944).

It is our opinion that the seismic crisis of 1999 marked the arrival of the magmatic fluids into the hydrothermal system, which caused the above-described geochemical variations that culminated in 2002. In support of this interpretation, Saccorotti et al. [2002] proposed that at least part of the seismic activity at Vesuvius is triggered by diffusion of a pressure front, possibly associated with the degassing process of the hydrothermal system.

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